

# Catalytic Materials in Green-Fuels Synthesis

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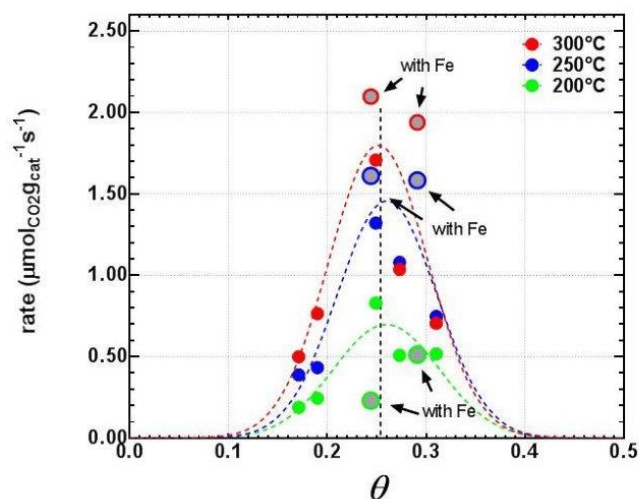
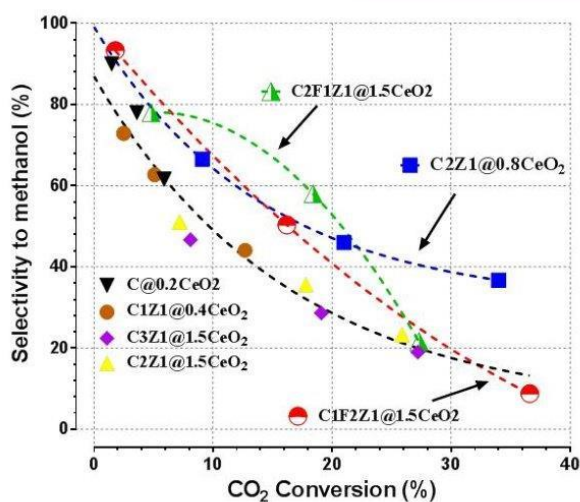
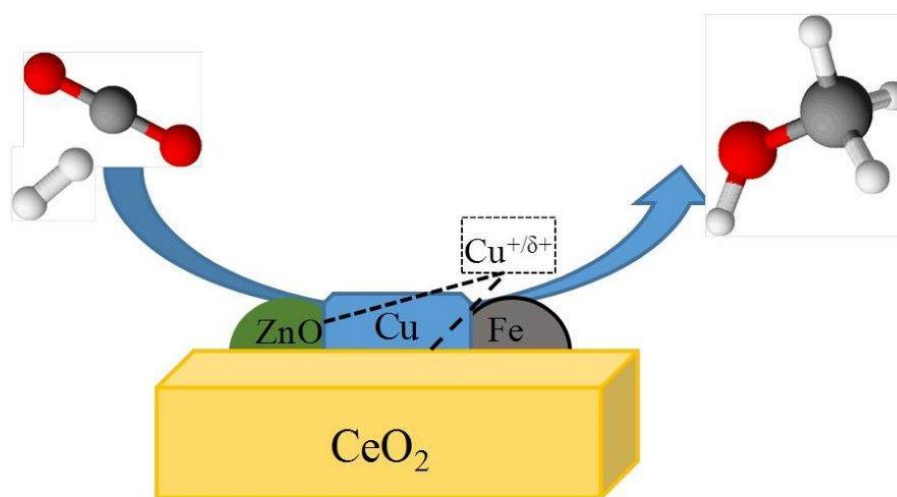
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## Graphical Abstract



## Abstract

Current prospective in the liquid fuels synthesis is prefiguring a greater integration of eco-friendly technologies based on the use of “non-fossil” hydrogen and CO<sub>2</sub>. Differently from fossil fuels, the “total-green-fuel” does not give rise to extra CO<sub>2</sub> emission in atmosphere, because its combustion generates a CO<sub>2</sub> release equal to that necessary for its manufacturing. Although the efforts to limit its presence in the atmosphere (CCUS technologies), CO<sub>2</sub> it is assuming an even more important and strategic role in the energy field, as well as in the synthesis of industrial relevant products and chemicals. As proof, the most important industry companies, such as Haldor-Topsoe, are at the fore front in the use of CO<sub>2</sub> as carbon feedstock for power-to-gas technologies, while other international petrol industries, as the Italian ENI, are redesigning the lines of production in a “greener” vision, by the development of new hydrogenation processes, such as ENI Ecofining<sup>™</sup>, for the green-fuels synthesis, strategically prefiguring the production of ultra-pure CO<sub>2</sub> as industrial practice. In this scenario, the development of more efficient catalytic materials for an effective hydrogenation of CO<sub>2</sub> appears of great interest. Therefore, the activity and selectivity pathways of MO<sub>x</sub>@CeO<sub>2</sub> (i.e. M=Cu, Fe and Zn) catalytic materials have been assessed in the CO<sub>2</sub> hydrogenation reactions, going insight on reaction mechanism and shedding light on the structure-activity relationships. In particular, a higher copper content leads to the sintering of the catalytic surface, affecting catalyst morphology and activity, although, the partial replacement of cerium with iron seems to slightly favour the exposure of the metallic surface area of copper, resulting almost unchanged the total surface area of catalyst. The catalytic findings point to the synergic catalytic action of copper-to-zinc and copper-to-cerium, reflecting structural and electronic effects, driving to a favourable Cu<sup>0</sup>/Cu<sup>+</sup> ratio and stabilizing Cu<sup>+</sup> sites when Cu<sup>0</sup>-O-Cu<sup>+</sup> species across CeO<sub>2</sub> interface.

**Keywords:** Advanced catalytic materials, CO<sub>2</sub>-to-fuels, renewable energy.

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## References

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